

**School of Engineering and Computer Science**  
**MECH 301: Thermodynamics**

<b>Catalog Data:</b>	<b>301 Thermodynamics</b> 3 Course Prerequisite: PHYSICS 201. Thermodynamic properties of matter, ideal and real gases, work and heat, first and second laws and their application to engineering systems. Typically offered Fall.	
<b>Class Schedule:</b>	Three 50-minute lecture sessions per week, for one semester.	
<b>Laboratory Schedule:</b>	None	
<b>Prerequisites by Course:</b>	PHYSICS 201	
<b>Prerequisites by Topic:</b>	<ol style="list-style-type: none"> <li>1. Basic knowledge of physics.</li> <li>2. Good understanding of calculus.</li> </ol>	
<b>Textbook:</b>	M.J. Moran, H.N. Shapiro, <i>Fundamentals of Engineering Thermodynamics</i>	
<b>Course Coordinator:</b>	Dr. Hua Tan	
<b>Course Objectives:</b>	<ol style="list-style-type: none"> <li>1. Ability to explain the basic principles of classical thermodynamics.</li> <li>2. Ability to find the thermodynamic properties of pure substances (using tables, charts, and ideal gas law) and apply them to thermodynamic analysis.</li> <li>3. Ability to identify, formulate, and solve engineering problems of closed and open systems involving heat and work interactions with the surroundings.</li> <li>4. Ability to apply the First and Second Laws of Thermodynamics to analyze thermal systems.</li> <li>5. Ability to determine properties of ideal gas mixtures and apply them to thermodynamic analysis.</li> <li>6. Ability to calculate properties of air-water mixtures and apply them to thermodynamic analysis.</li> </ol>	
<b>Topics Covered:</b>	<ol style="list-style-type: none"> <li>1. Thermodynamic concepts</li> <li>2. Closed systems – 1<sup>st</sup> Law</li> <li>3. Properties of pure substances</li> <li>4. Open systems – 1<sup>st</sup> Law</li> <li>5. Systems - 2<sup>nd</sup> Law</li> <li>6. Thermodynamic cycles</li> <li>7. Ideal gas mixtures</li> <li>8. Thermodynamics of state</li> </ol>	
<b>Lab Experiments and Activities:</b>	None	
<b>Course Outcomes:</b>	Students will be able to:	
	<b>Assessed for Student Outcomes</b>	<ol style="list-style-type: none"> <li>1-b. Classify thermodynamic problems, such as closed or open systems, depending on conditions or assumptions defined in engineering problems.</li> <li>4-a. Apply thermodynamic laws to explain engineering decisions for new energy resources.</li> </ol>

	<b>Other</b>	1-a. Review the fundamental conservation laws of mass, momentum, and energy. 1-c. Apply 1 <sup>st</sup> and 2 <sup>nd</sup> law of thermodynamics, thermodynamic cycles, ideal gas, and psychometrics to solve engineering processes and systems. 1-d. Apply mathematical principles to derive equations for conservation of mass, momentum, and energy for the thermodynamic problems.	
<b>Required or Elective Course:</b>	Required		
<b>Relationship of Course to Program:</b>	Meets: Educational Objectives <u>1, 3</u> Student Outcomes <u>1, 4</u>		
<b>Prepared by:</b>	Dr. Hua Tan	Date:	4/9/2018 (4/10/18 mb)
<b>Approved by USC:</b>	4/9/18		